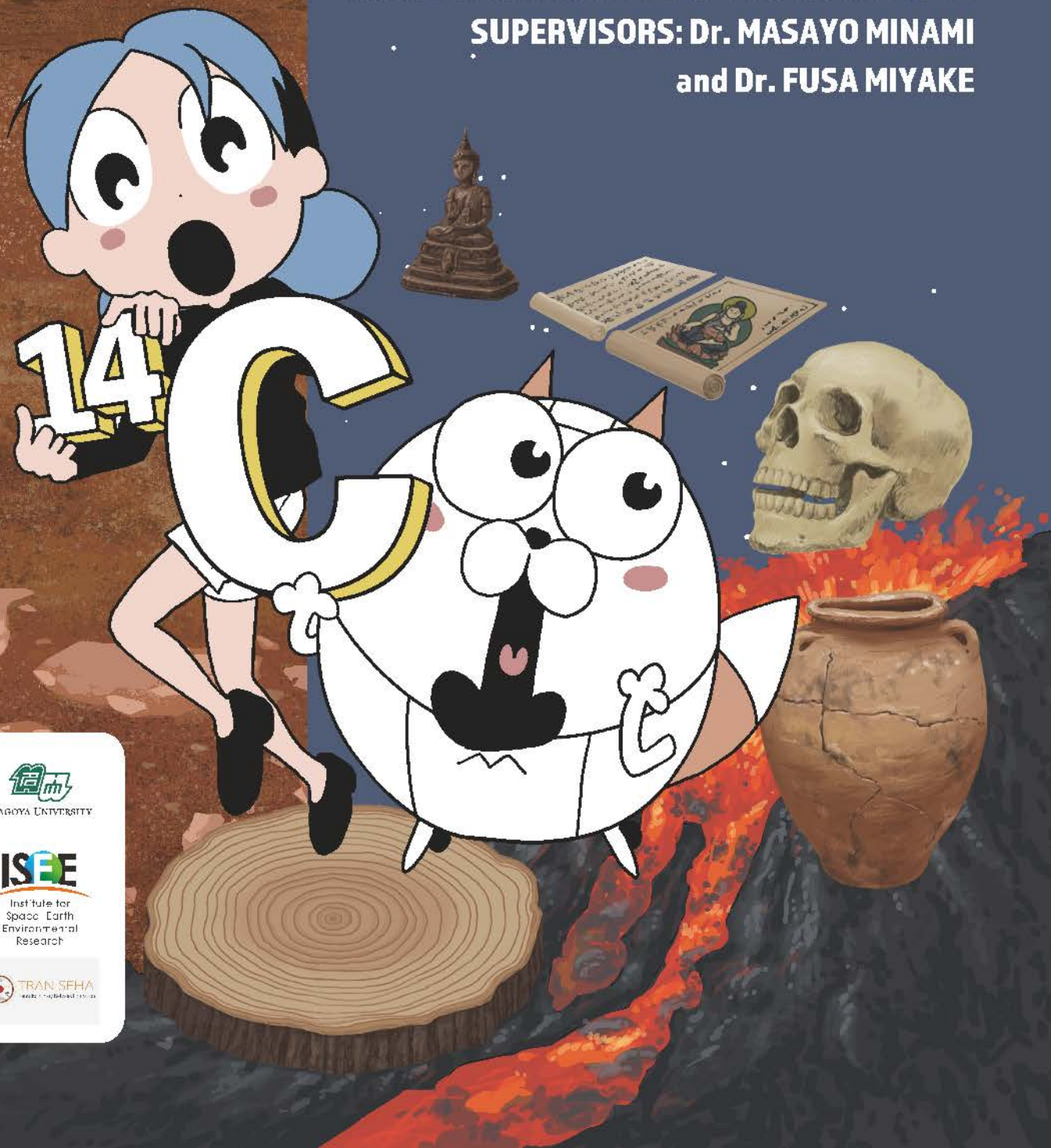


# What is $^{14}\text{C}$ ?!

MANGA: HAYANON+SCIENCE MANGA STUDIO

SUPERVISORS: Dr. MASAYO MINAMI

and Dr. FUSA MIYAKE



NAGOYA UNIVERSITY



Institute for  
Space, Earth  
Environmental  
Research



TRANSHEA  
earth & space research

Can we really learn about the past just by studying "carbon"? Carbon is all around us—in our bodies, in the food we eat, and in everyday objects. A special type of carbon called  $^{14}\text{C}$  (Carbon-14) is produced in atmosphere by cosmic rays from space. It is known to decay over time, with half of it disappearing every 5,730 years. This unique property enables scientists to determine the age of various things, and it is useful in fields such as Earth science, space science, archaeology, and anthropology. Researchers have even used it to study past changes in the Earth's environment and Sun's activity for parallelism. In this comic, let's learn all about  $^{14}\text{C}$  with Mirubo, Moru-chan, and professors!

## What is $^{14}\text{C}$ ?!



### Dr. MASAYO MINAMI

Professor,  
Office for the Promotion of Transdisciplinary  
Network / Division for Chronological Research,  
ISEE, Nagoya University.

### MORU-CHAN

She's an energetic elementary  
schooler who loves science—  
a bit of a show-off,  
but always full of curiosity!

$^{14}\text{C}$

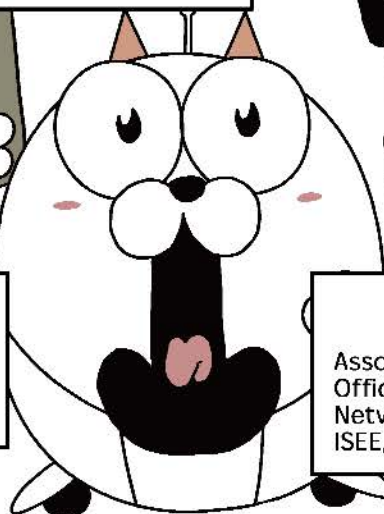


### Dr. FUSA MIYAKE

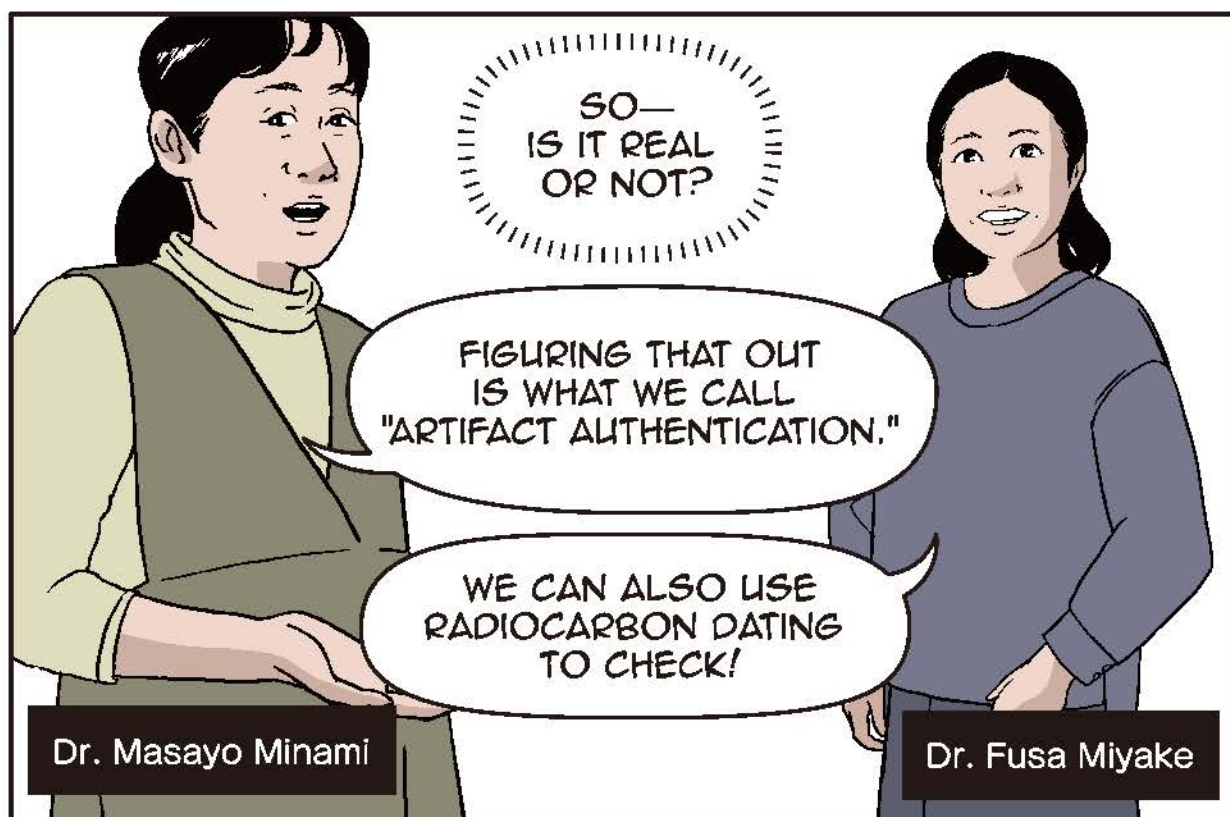
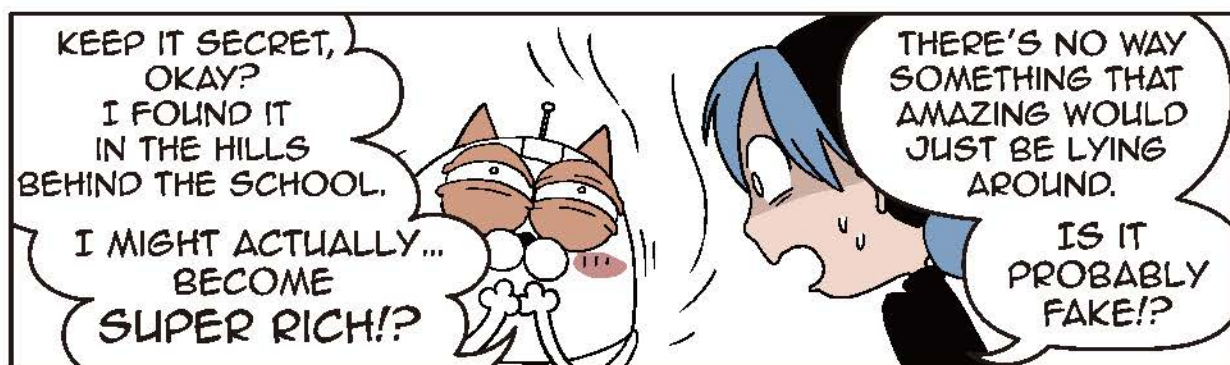
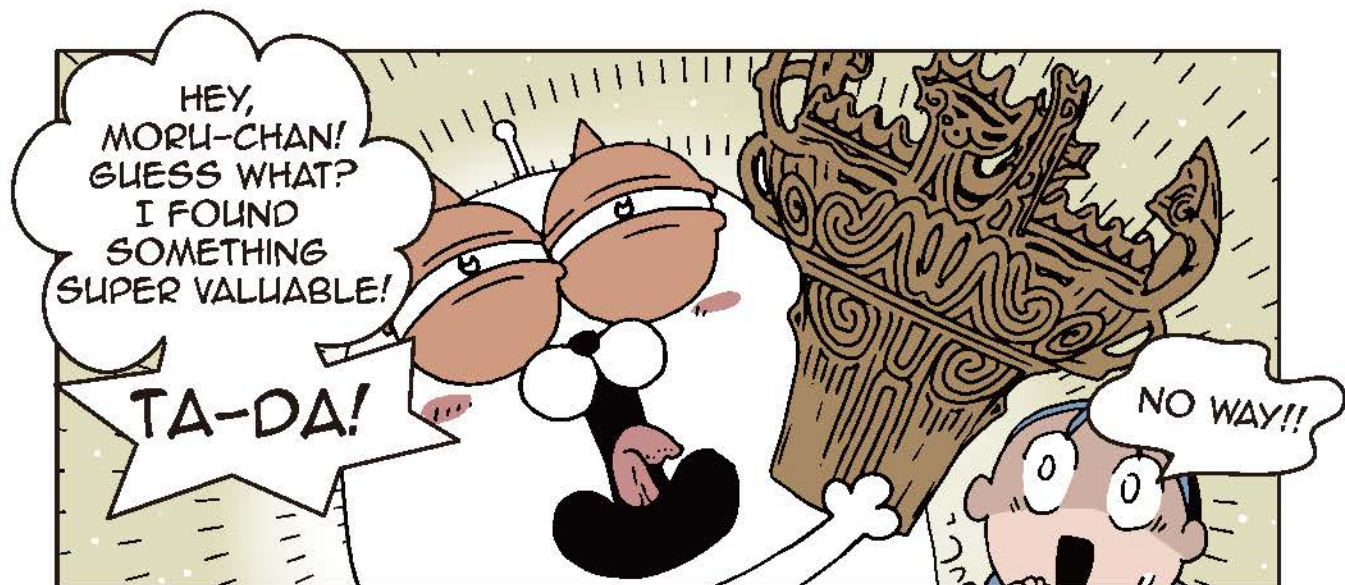
Associate professor,  
Office for the Promotion of Transdisciplinary  
Network / Division for Cosmic-ray Research,  
ISEE, Nagoya University.

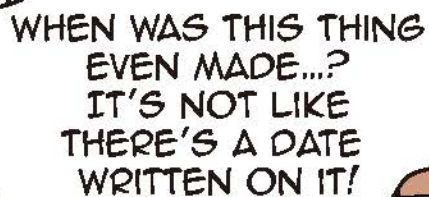
### MIRUBO

A cutting-edge robot dog!  
Totally obsessed with  
grilled meat.

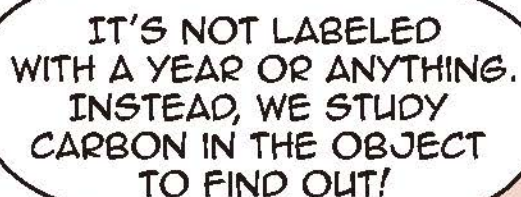








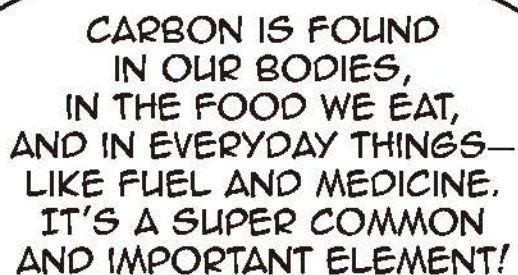
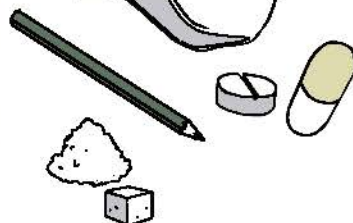
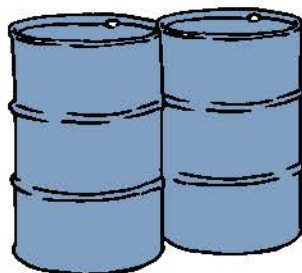
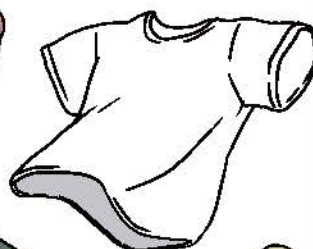
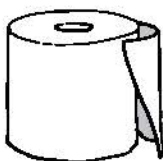
WHEN WAS THIS THING  
EVEN MADE...?  
IT'S NOT LIKE  
THERE'S A DATE  
WRITTEN ON IT!



IT'S NOT LABELED  
WITH A YEAR OR ANYTHING.  
INSTEAD, WE STUDY  
CARBON IN THE OBJECT  
TO FIND OUT!



CARBON?



CARBON IS FOUND  
IN OUR BODIES,  
IN THE FOOD WE EAT,  
AND IN EVERYDAY THINGS—  
LIKE FUEL AND MEDICINE.  
IT'S A SUPER COMMON  
AND IMPORTANT ELEMENT!

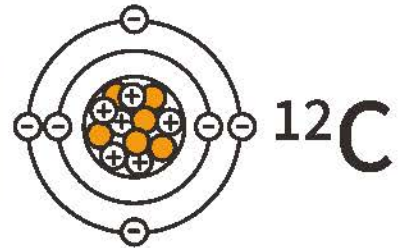


CARBON ATOMS HAVE A NUCLEUS WITH 6 PROTONS AT THE CENTER.

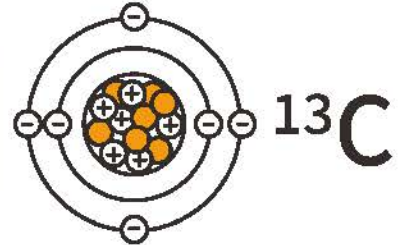
NORMALLY, THEY ALSO HAVE 6 NEUTRONS, BUT SOME HAVE 7 OR EVEN 8.

ATOMS WITH THE SAME NUMBER OF PROTONS BUT DIFFERENT NUMBERS OF NEUTRONS ARE CALLED ISOTOPES.

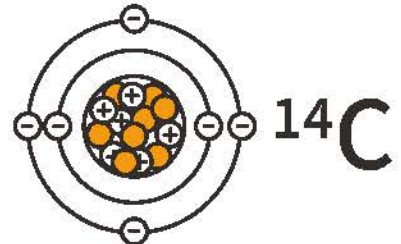
6 Protons +  
6 Neutrons ●



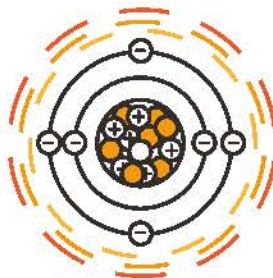
6 Protons +  
7 Neutrons ●



6 Protons +  
8 Neutrons ●



ONE OF THESE ISOTOPES,  $^{14}\text{C}$ , OR CARBON-14, HAS EXTRA NEUTRONS, MAKING IT UNSTABLE.



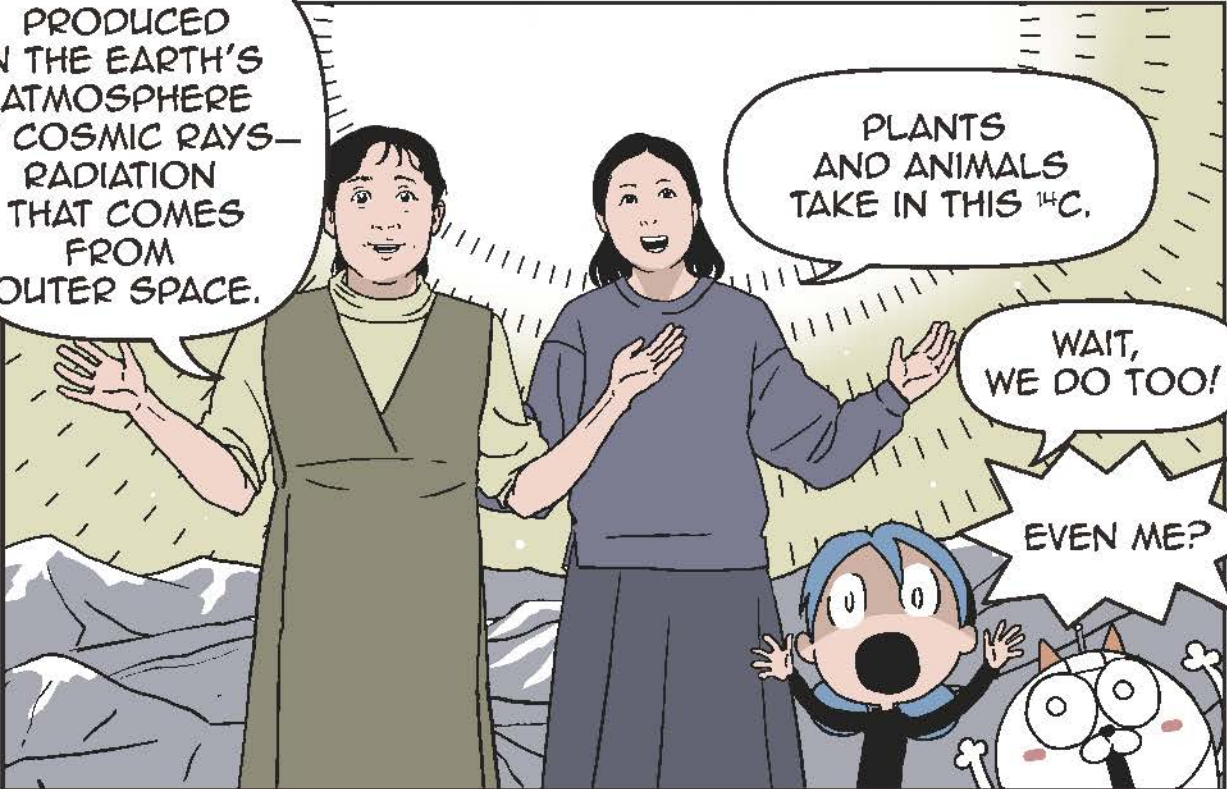
$^{14}\text{C}$  UNDERGOES RADIOACTIVE DECAY...

IT DECAYS...

AND DECAYS...

UNTIL HALF OF IT IS GONE—  
AFTER 5,730 YEARS!

THIS IS CALLED THE "HALF-LIFE."



$^{14}\text{C}$  IS  
PRODUCED  
IN THE EARTH'S  
ATMOSPHERE  
BY COSMIC RAYS—  
RADIATION  
THAT COMES  
FROM  
OUTER SPACE.

PLANTS  
AND ANIMALS  
TAKE IN THIS  $^{14}\text{C}$ .

WAIT,  
WE DO TOO!

EVEN ME?



ONCE  
A LIVING THING DIES,  
IT STOPS TAKING IN  
ANY NEW  $^{14}\text{C}$ .

BY MEASURING  
THE AMOUNT OF  $^{14}\text{C}$   
LEFT IN THE BODY—

WE CAN FIGURE OUT  
THE AGE OF AN OBJECT!

THIS METHOD IS CALLED  
"RADIOCARBON DATING,"  
AND IT'S WIDELY USED  
IN ARCHAEOLOGY,  
GEOLOGY, AND  
MANY OTHER FIELDS.

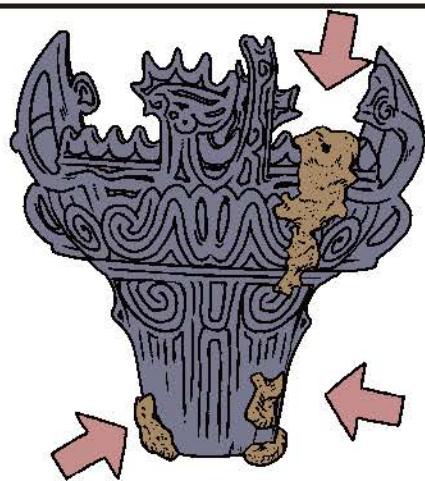


PLEASE  
CHECK OUT  
THIS POTTERY  
I FOUND,  
PROFESSOR!!!

WAIT—  
ARE THERE  
MORE NOW!?

WE'RE NOT TESTING  
THE POTTERY ITSELF—  
WE LOOK AT CARBON  
IN THE STUFF STUCK TO IT.

LIKE, IF SOMEONE  
USED THAT POT  
TO COOK FOOD LONG AGO,  
WE CAN ANALYZE  
THE BURNED MATERIAL  
LEFT BEHIND TO FIGURE OUT  
WHEN IT WAS USED.

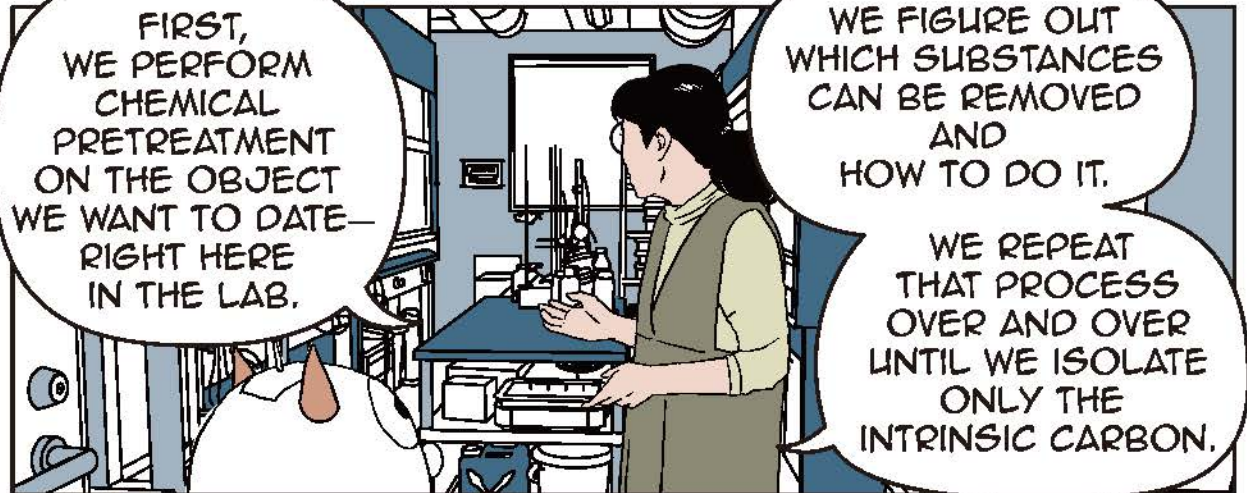


BUT OVER TIME,  
THOSE THINGS  
GET CONTAMINATED  
BY ALL TYPES OF MATERIALS—  
SO WE NEED TO CAREFULLY  
REMOVE CARBON  
CONTAMINANTS  
TO EXTRACT  
INTRINSIC CARBON.

THIS PROCESS  
IS CALLED  
CHEMICAL PRETREATMENT  
THROUGH WHICH  
WE SEPARATE MATERIALS  
AND STUDY THEM.

AND PROFESSOR  
MINAMI IS AN EXPERT  
IN THIS FIELD!

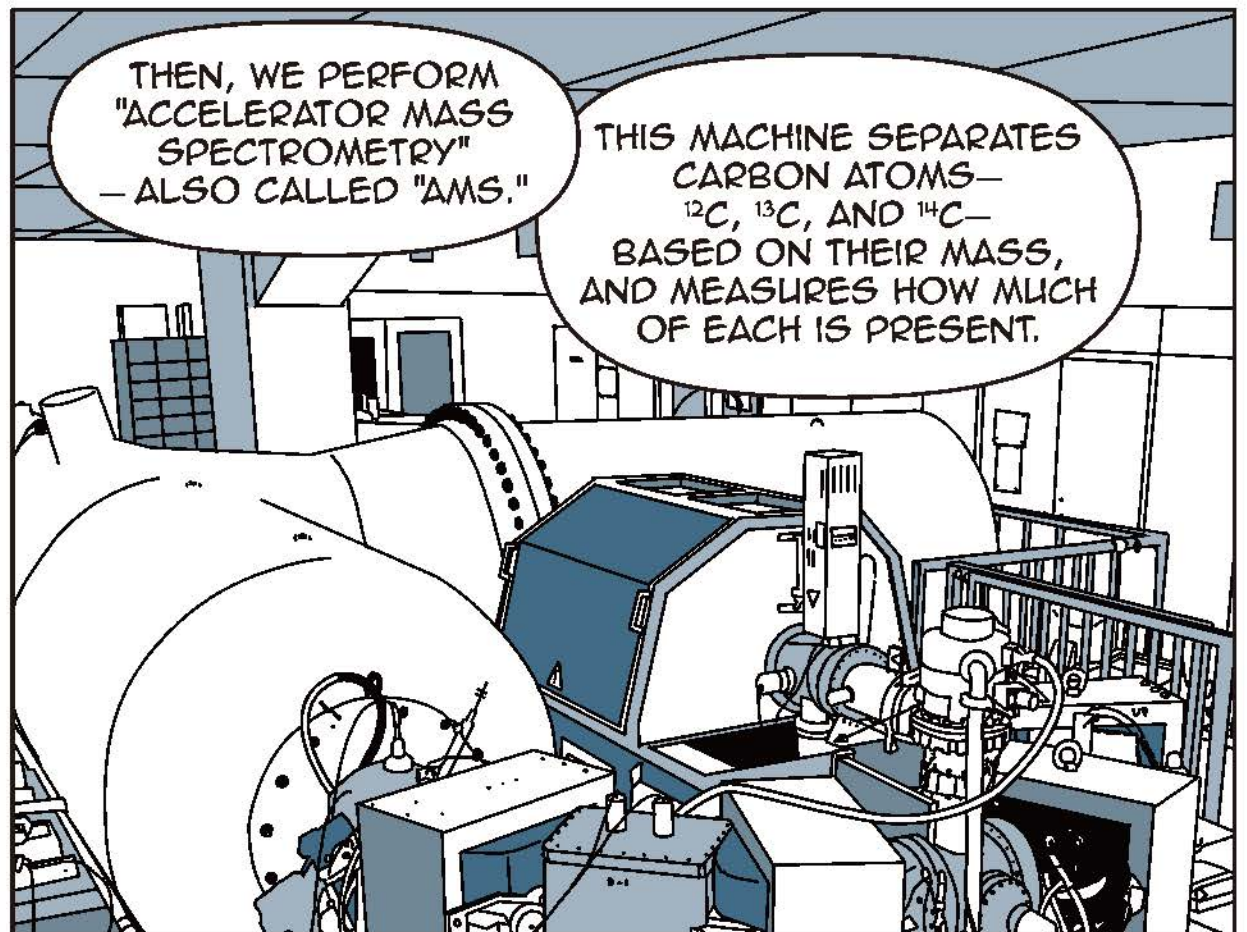




FIRST,  
WE PERFORM  
CHEMICAL  
PRETREATMENT  
ON THE OBJECT  
WE WANT TO DATE—  
RIGHT HERE  
IN THE LAB.


WE FIGURE OUT  
WHICH SUBSTANCES  
CAN BE REMOVED  
AND  
HOW TO DO IT.

WE REPEAT  
THAT PROCESS  
OVER AND OVER  
UNTIL WE ISOLATE  
ONLY THE  
INTRINSIC CARBON.



THEN, WE PERFORM  
"ACCELERATOR MASS  
SPECTROMETRY"  
— ALSO CALLED "AMS."

THIS MACHINE SEPARATES  
CARBON ATOMS—  
 $^{12}\text{C}$ ,  $^{13}\text{C}$ , AND  $^{14}\text{C}$ —  
BASED ON THEIR MASS,  
AND MEASURES HOW MUCH  
OF EACH IS PRESENT.



THAT WAY, WE CAN GET  
A CLEAR NUMBER  
SHOWING HOW MUCH  $^{14}\text{C}$   
IS STILL LEFT  
IN THE SAMPLE.



THIS ANALYSIS TELLS US HOW MUCH  $^{14}\text{C}$  IS LEFT IN THE SAMPLE.

BUT JUST KNOWING THE HALF-LIFE DOESN'T MEAN WE CAN SAY, "THIS MUCH  $^{14}\text{C}$  = THIS MANY YEARS"—NOT EXACTLY.

THERE CAN BE A PRETTY BIG GAP BETWEEN THE MEASURED  $^{14}\text{C}$  AND THE ACTUAL CALENDAR YEAR.

WHY? BECAUSE—

THE AMOUNT OF  $^{14}\text{C}$  CHANGES DEPENDING ON THINGS LIKE SOLAR ACTIVITY AND THE EARTH'S MAGNETIC FIELD.

IT ALSO VARIES BY LOCATION.

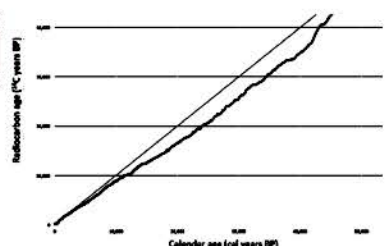
It differs between the Northern and Southern Hemispheres and between land and ocean.

EVEN HUMAN ACTIVITIES—BURNING FOSSIL FUELS OR DOING NUCLEAR TESTS—CAN CHANGE THE AMOUNT OF  $^{14}\text{C}$ .

THAT'S WHY WE USE SOMETHING CALLED A CALIBRATION CURVE TO CORRECT THOSE DIFFERENCES! THIS CURVE IS SUPER IMPORTANT WHEN DOING RESEARCH WITH  $^{14}\text{C}$ .

International Calibration curve: IntCal  
Originally proposed in 1998,  
and updated in 2004, 2009, 2013, and 2020.

Calibration curve





THE SCIENCE OF  
DATING THINGS—  
FIGURING OUT THE TIME  
SOMETHING IS FROM—  
IS USEFUL IN MANY FIELDS.

BY STUDYING  
LAYERS OF EARTH,  
WE CAN FIND OUT WHEN  
ANCIENT  
TSUNAMIS HAPPENED!

IN ARCHAEOLOGY,  
WE CAN DATE ARTIFACTS  
BY ANALYZING  
BURNED MATERIAL  
ON POTTERY OR  
EVEN THE CARBON IN  
OLD DOCUMENTS.

IN ONE CASE,  
RADIOCARBON DATING PROVED  
AN ANCIENT DOCUMENT  
WAS REAL—AND IT BECAME  
A NATIONAL TREASURE!

IN THE ART WORLD,  
IT HELPS IDENTIFY  
WHAT ERA  
A PIECE WAS MADE IN.

AND IN  
FORENSIC SCIENCE,  
IT CAN EVEN DETERMINE  
WHEN A PERSON DIED.

SOUNDS LIKE  
SOMETHING  
STRAIGHT OUT OF  
A CRIME DRAMA!





$^{14}\text{C}$  ISN'T JUST  
USEFUL FOR  
DATING  
ANCIENT THINGS.

FROM THE  $^{14}\text{C}$   
IN TREE RINGS,  
WE CAN LEARN ABOUT  
PAST SOLAR ACTIVITY!

WHEN A MASSIVE  
EXPLOSION HAPPENS ON  
THE SUN'S SURFACE,  
IT SENDS OUT  
HIGH-ENERGY PARTICLES,  
MAKING THE AMOUNT OF  $^{14}\text{C}$   
ON THE EARTH GO UP!

Solar flares and  
coronal mass ejections (CMEs)  
release high-energy particles  
that reach the Earth.

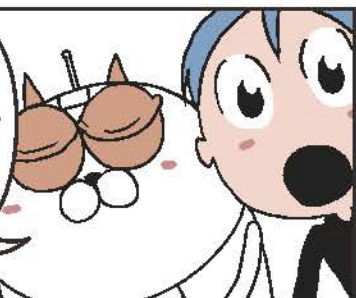
WE STUDIED  
YAKUSUGI\* TREE RINGS  
AND FOUND EVIDENCE OF  
HUGE SOLAR FLARES  
THAT OCCURRED IN  
THE YEARS 774 AND 993!

\*ancient cedar trees  
from Yakushima Island

HUGE SOLAR FLARES  
CAUSE SHARP SPIKES  
IN THE AMOUNT OF  $^{14}\text{C}$ .

SCIENTISTS WORLDWIDE  
ARE NOW PAYING ATTENTION  
TO THESE  $^{14}\text{C}$  SPIKES,  
BECAUSE THEY CAN  
HELP PINPOINT EXACT YEARS  
WITH AMAZING ACCURACY!





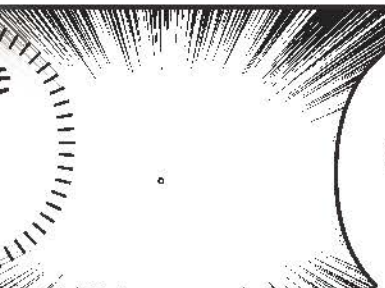
AH, I GET IT!  
SO BY STUDYING  $^{14}\text{C}$ ,  
WE CAN LEARN  
ALL KINDS OF THINGS  
ABOUT THE PAST!

LIKE,  
WHAT TIME PERIOD  
SOMETHING'S FROM—  
HOW FAR BACK  
CAN WE  
ACTUALLY GO?



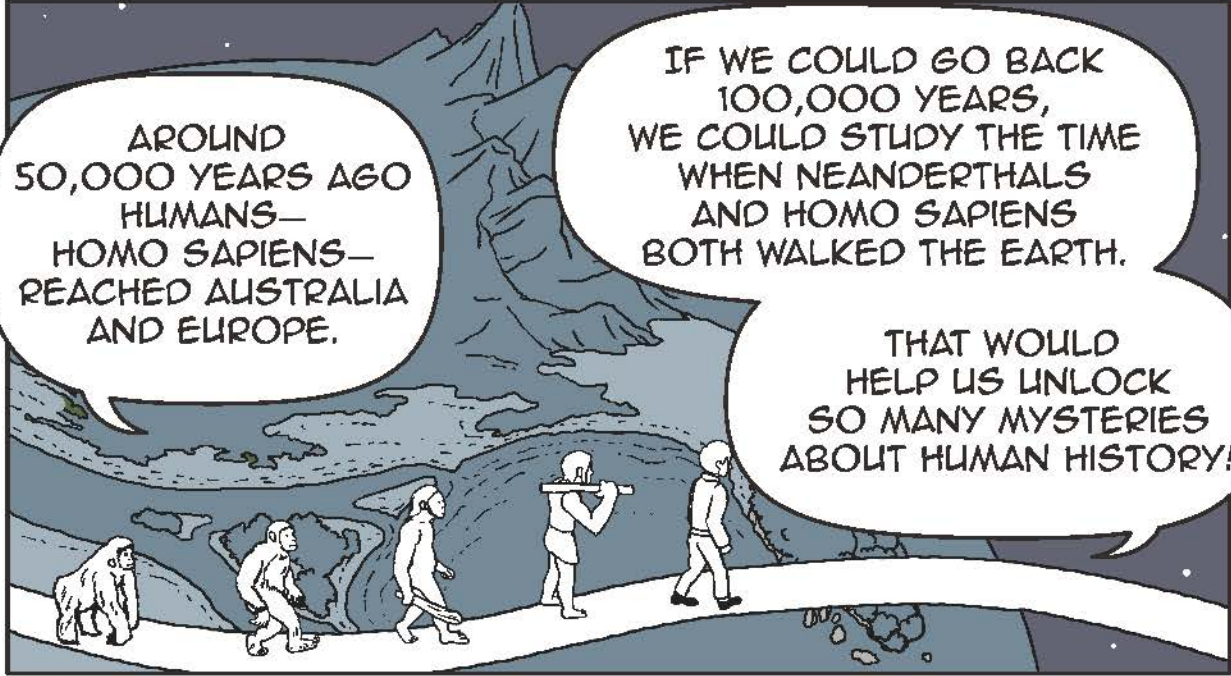
IT'S SAID WE CAN  
MEASURE THINGS  
UP TO ABOUT  
10 TIMES  
THE HALF-LIFE  
OF  $^{14}\text{C}$ .

SINCE THE  
HALF-LIFE OF  $^{14}\text{C}$   
IS 5,730 YEARS,  
THAT MEANS  
WE CAN GO BACK  
AROUND  
55,000 TO  
60,000 YEARS.



I WANT TO BE ABLE  
TO STUDY THINGS  
FROM EVEN  
FURTHER BACK  
IN TIME!

IF WE COULD  
MEASURE DATES  
EVEN EARLIER THAN  
THAT WITH  $^{14}\text{C}$ ,  
THAT WOULD BE  
AMAZING!



AROUND  
50,000 YEARS AGO  
HUMANS—  
HOMO SAPIENS—  
REACHED AUSTRALIA  
AND EUROPE.

IF WE COULD GO BACK  
100,000 YEARS,  
WE COULD STUDY THE TIME  
WHEN NEANDERTHALS  
AND HOMO SAPIENS  
BOTH WALKED THE EARTH.

THAT WOULD  
HELP US UNLOCK  
SO MANY MYSTERIES  
ABOUT HUMAN HISTORY!



SO—  
THE POTTERY  
MIRUBO BROUGHT IN  
TURNED OUT  
TO BE REAL!

IT LOOKS LIKE  
THE FLAME-STYLE  
POTTERY\* FROM ABOUT  
5,000 YEARS AGO!

YES!!

I'VE GOT LOTS  
MORE TREASURES, TOO!  
I WANNA STUDY MORE  
SO I CAN FIGURE  
THESE THINGS OUT  
ON MY OWN SOMEDAY!

WAIT... WHERE  
DID HE EVEN—

THROUGH  $^{14}\text{C}$ ,  
WE CAN LEARN  
ABOUT THE EARTH,  
THE SUN,  
AND HUMAN HISTORY.  
BUT THERE'S  
STILL SO MUCH  
WE DON'T KNOW.

IF YOU FEEL CURIOUS  
AND WANT TO  
LEARN MORE,  
TRY READING BOOKS  
AND STUDYING!

AND IF YOU'RE  
STILL CURIOUS—  
GO STUDY IT  
IN COLLEGE!

ALL RIGHT!  
LET'S DO THIS!!

# What is $^{14}\text{C}$ ?!



Mirubo :  $^{14}\text{C}$  is so useful!  
What should I try dating next?



Moru-chan : Professor!  
Can we use  $^{14}\text{C}$  to date  
dinosaur fossils?



Dr. Minami : Great question!  
Radiocarbon dating works for  
things that are up to about  
50,000 years old. However,  
dinosaurs lived over 65 million  
years ago, so this method  
doesn't work for them.



Dr. Miyake : When we date  
dinosaur fossils, we usually  
look at rock layers or minerals  
around fossils— not fossils  
themselves. We often use  
methods like uranium-lead  
technique.



Mirubo : So, it sounds like  
there are lots of dating  
methods besides  $^{14}\text{C}$ !



Moru-chan : I remember  
learning that you can date  
things with an age of about 10  
times the half-life of an isotope  
using that isotope. So, if we  
use elements with a longer  
half-life, can we date stuff from  
way earlier?



Mirubo : (Hmm, sharp  
question!) I was just about to  
ask the same thing!



Dr. Minami : Exactly.  
For example, beryllium-10  
( $^{10}\text{Be}$ ) has a half-life of about  
1.4 million years. So, we can  
use it to date samples that are  
5 million years old. It's actually  
used to study the age of soils,  
sediments, and ice sheets.



Mirubo : Whoa—awesome!  
All right, let's try using that  
method right now!



Dr. Miyake : Well, we can only  
use it on things that actually  
contain beryllium.



Dr. Minami : Meanwhile,  
carbon is found in almost all  
living things and artifacts.  
That's why  $^{14}\text{C}$  is so widely  
used across many different  
fields.



Mirubo :  $^{14}\text{C}$  is the best!  
It makes me wanna analyze  
everything!



Dr. Minami :  $^{14}\text{C}$  dating often  
pops up on TV shows like  
dramas and variety shows  
— so lots of people have  
heard of it.



Dr. Miyake : But, isn't  $^{14}\text{C}$  just  
for dating objects?  
Because  $^{14}\text{C}$  is produced by  
cosmic rays, variations in  
cosmic-ray intensity also  
affect the amount of  $^{14}\text{C}$ .



Moru-chan : How do you get  
carbon from tree rings  
specifically?





Dr. Minami : Great question!  
The thing we're testing is called a sample. It's super important to treat the sample properly—to remove any contaminants through chemical processing.



Mirubo : So, it's like cooking, right? You prep your ingredients carefully to bring out the best flavor!



Dr. Minami : Exactly! For example, with tree rings, we remove any contaminants from the wood and purify just carbon we need—so, we can measure it accurately. How well we prepare the sample makes a huge difference in how precise the results are.



Moru-chan : Who usually does this kind of research?



Dr. Miyake : At ISEE, Nagoya University, we are applying the knowledge and techniques of space and Earth environmental science researchers to studies in the fields of history and archaeology. We also work with specialists in precision measurement, archaeologists, and data analysts. Everyone contributes their knowledge and skills to make new discoveries together!



Mirubo : I wanna do  $^{14}\text{C}$  research myself someday! All right—maybe I'll become a scientist too!

What kind of research can we do with  $^{14}\text{C}$ ?  
If you come up with a great idea, you may just become a scientist someday!





## Institute for Space-Earth Environmental Research, Nagoya University

In 2015, three major research centers at Nagoya University —Solar-Terrestrial Environment Laboratory, Hydrospheric-Atmospheric Research Center, and the Center for Chronological Research— were merged to form the Institute for Space-Earth Environmental Research. This integration made it possible to conduct more interdisciplinary research, connecting different scientific fields in new ways. The ISEE is the only institute in Japan that bridges space science and Earth science, working on global environmental challenges and the sustainable use of outer space through international collaborations.

<https://www.isee.nagoya-u.ac.jp/>



TRAN SEHA  
地球ネットワーク形成

## TranSEHA

NEXT Promotion of Development of a Joint Usage/Research System Project:  
Coalition of Universities for Research Excellence Program (CURE)

In fiscal year 2024, Institute for Space-Earth Environmental Research (ISEE) was selected for a major new research program. As a part of this initiative, ISEE is partnering with five institutions: the National Museum of Japanese History, Yamagata University's Center for Advanced Accelerator Mass Spectrometry, Kyushu University's Center for Asian Archaeological Research, the Joint Support-Center for Data Science Research, and Nagoya University's Center for Digital Humanities and Social Sciences.

Together, these institutions are building a new transdisciplinary research network that brings together Space-Earth environmental science with history and archaeology in innovative and meaningful ways.

Through this program, researchers will evaluate the potential impacts of extreme solar storms on modern civilization and advance history and archaeology by developing new, highly accurate methods of chronological dating.

The project also aims to contribute to the creation of a sustainable, evolving society that can expand into space, while improving preparedness for space weather disasters, earthquakes, and volcanoes.

At the same time, it seeks to foster the next generation of globally minded researchers who will lead future interdisciplinary science. The program is planned to run for up to 10 years, beginning in fiscal year 2024.



<https://transeha.isee.nagoya-u.ac.jp/en>



## Hayanon

Science manga artist, born in 1975. Graduated from the Department of Physics, Faculty of Science, University of the Ryukyus, with a B.A. (Science). Completed M.A. (Education) in English Education from the Graduate, Graduate School of Education, Chiba University. Her representative works include "GoGo! MIRUBO" (Kodomo no Kagaku), "Fantastic R&D" (Nikkan Kogyo Shimbun), "Learning Earth Science with MIRUBO" (NASA), and "International Science Olympiad Manga" (Japan Science and Technology Agency). She is a representative of Science Manga Studio, a PR for research business.

## Science Manga Studio, Japan

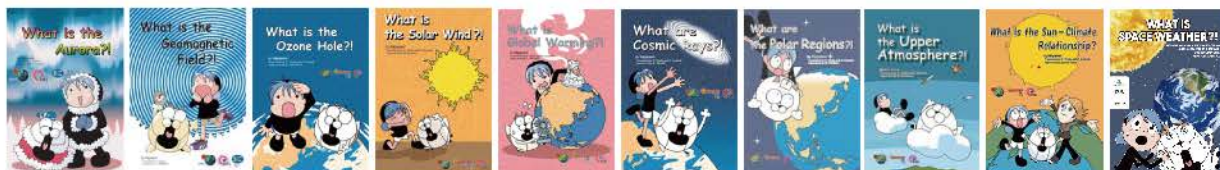
The studio produces manga introducing research in all fields of science and humanities, cover art for academic journals, and provides illustrations explaining the results of research institutions' presentations. Through these works, the studio fosters young science illustrators. <https://www.sciencemanga.com/>

### Science illustrators in the production

Ms. Shell, Mochizuki Ami, Hakushi Jaco, and Akiba from Science Manga Studio.

## What Is...?! Series

This is a research introduction Manga series started in 2002. You can learn about the study of the Sun and Earth with Sensei, Mirubo, and Mol.



<https://www.isee.nagoya-u.ac.jp/en/outreach.html>

ISEE MIRUBO

Q SEARCH

PDFs are distributed free of charge. The original editions were written in Japanese, and translations are available in English and other languages. <https://scostep.org/space-science-comic-books/>